



2024

at a glance



Advanced
Research Center
Chemical Building
Blocks Consortium



Table of contents



Timeline	3
In the Spotlights	
<i>Jens Tolboom</i>	4
<i>Marina Karsakova</i>	5
Our Story	6
Our Research	7
In the Spotlights	
<i>Işıl Yeşil Gür</i>	9
<i>Bas den Hartigh</i>	10
Outreach	
Events	11
In the media	12
Education	13
In the Spotlights	
<i>Alexandra Matei</i>	14
<i>Jonas Gans</i>	15
Awards, Grants and other Honours	16
Our People	17
Our New Researchers	19
Our Alumni	21
Contact Details	22

Timeline 2024



23-24 January
NWO Physics



9 May
Refinery of the Future in Nature Magazine

10-12 June
ARC CBBC Summer School



22 August
Strategic partnership CAPTURE announced

6-13 September
ARC CBBC participates in Innovation mission China



23-24 October
ARC CBBC visit to BASF Ludwigshafen



3-4 December
NWO CHAINS

4 March
NCCC



6 June
Circular Biobased Product Symposium

24-27 June
Coatings Science International (CoSI)



29-30 August
NextGenChem@NL



15 October
ARC CBBC Community Event



21 November
ChemistryNL Future Festival

3 December
Launch ARC CBBC Edition C2W



Jens Tolboom (UvA)

When it comes to paint, we're quite spoiled: we expect it to be fluid when applied and to dry as quickly as possible afterwards. To most, this is perfectly normal. However, what many people don't realise is that this process doesn't just happen on its own – there is a lot of chemistry behind it. How do you prevent your paint from drying too soon, and thus a skin from forming in the tin, while ensuring it dries quickly once applied to the wall?

Conventional paints contain substances that prevent a skin from forming in the tin and others that accelerate the drying process immediately after application. However, these substances may be harmful, so it would be ideal if their use could be avoided. Jens Tolboom is working on a solution for this at the University of Amsterdam. He is developing a sustainable catalyst that can be added to paint, using a very clever trick: a light switch.

✍ Jens Tolboom receives Best Young Scientist Contribution Award at ETCC 2024

A light-sensitive switch is a promising concept for drying paint: still in the tin, it's kept in the dark, while once applied to the wall, it's exposed to an abundance of light. If paint were to dry as a reaction to light, it would have precisely the properties you want. Jens is designing iron-based catalysts that accelerate drying when exposed to light. Is there no light? Then the catalyst remains inactive, and the paint won't dry. Is there enough light? The catalyst becomes active, and the paint will dry. Doesn't that sound ideal?

Still, there are some obstacles to overcome. What about dark-coloured paint, such as black or dark blue paint? The pigments that give the paint its colour block the light from

entering it. And if the light fails to reach the catalysts, they will remain inactive, and the paint will not dry. Jens is looking for a solution for specifically that problem. He is doing this by designing catalysts that use the specific part of light that is nevertheless able to pass through pigments, and which will be activated even in dark-coloured paints.

The addition of such catalysts could make paint more sustainable and help avoid the use of potentially harmful substances. We are eagerly looking forward to the day when the product of Jens' research can actually be found in tins of paint. Jens was conferred the Best Young Scientist Award at the 2024 ETCC conference, so he's well on his way!

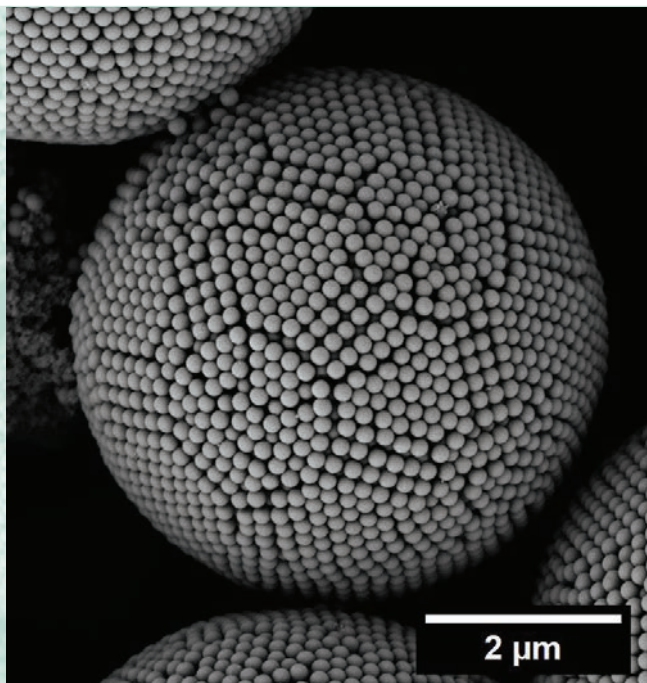


IN THE SPOTLIGHTS

Marina Karsakova (RUG)

We all know the disappointment when our brightly coloured items start to lose their charm. Why does this happen? Most often, it's because the colour fades, losing the vibrant appeal it had when it was new. It's a shame, especially when they still work perfectly fine! But what if we could create colours that don't fade, keeping our items as vibrant as the day we bought them? This is exactly what Marina Karsakova is investigating in her research at the University of Groningen.

Many colours fade over time because the pigments or dyes in paints and fabrics break down under the influence of light. Take, for example, an old red car left outside for years—its once-brilliant paint becomes dull. This happens because the molecules in the paint, called chemical pigments, absorb light and undergo structural changes. These changes alter the molecules so much that they effectively transform into a different substance with a different colour. Therefore, over time, the bright red turns faded and lacklustre.



This fading doesn't just affect cars or clothing—it happens to many products, which are often replaced simply because they no longer look appealing, even though their functionality remains intact. Marina's work aims to solve this problem by developing "structural pigments" that don't fade, even when exposed to light.

So, what makes structural pigments special? Unlike traditional chemical pigments, structural pigments create colour through their physical structure rather than through chemical compounds that can break down. These structural pigments, as pioneered by the Van Blaaderen Group at Utrecht University, are made from tiny, ordered spheres, often using inexpensive and durable materials like sand. These spheres interact with light in precise ways to produce vibrant, stable colours that don't fade. Amazingly, using just one material, it's possible

Photo left: Marina's structural pigments

Photo below: Marina receiving the Poster Prize at the ARC CBBC Community Event

- ✍ Marina Karsakova in C2W ARC CBBC special
- ✍ ARC CBBC Community Event 2024

to create all the colours of the rainbow, from blue to red. Although the individual spheres are far too small to see with the naked eye, their effect is impossible to miss—they ensure colours stay brilliant over time.

Marina is using this method to design sustainable pigments that last longer and retain their colour intensity. The research she conducted in 2024 also earned her two poster prizes in one and the same month, both at the ARC CBBC Community Event and the Münster Symposium on Intelligent Matter. And, as the icing on the cake, her research was given a prominent place in C2W's special ARC CBBC edition!





Our Story

As illustrated by our slogan 'Reinventing chemistry together', ARC CBBC has been serving as a bridge between the academic world and the chemical industry for eight years now. This public-private partnership brings out the best in both worlds, in which academic need-to-know and industrial need-to-have inspire one another other think beyond the boundaries of individual disciplines. With over 60 projects, we are well on the road to achieving our goals – but still have a distance to travel.

Our first projects were launched in 2016. They were set up in conjunction with our hub universities (Utrecht University, the University of Groningen and Eindhoven University of Technology) and our founding industrial partners (AkzoNobel, Shell, BASF, Nouryon and Nobian). The first three multilateral projects we launched were Coatings, Small Molecule Activation and Fundamentals of Catalysis. Aside from these, multiple projects were initiated by the above organizations as well as associate partners throughout the Netherlands, all with a view to optimizing the exchange of expertise.

In the meantime, eight years have passed. Our first projects have largely been concluded, but that does not mean that we have reached our goal. The topics we are addressing are

still far from resolved. This is why we have launched new multilateral projects that continue where their predecessors left off: Smart Coatings, Methane Pyrolysis and CO₂ Conversion.

Our purpose is to rethink the design of the chemical building blocks that make up the products of our everyday lives, and the convenience they bring us. We investigate manufacturing routes and the use of chemical feedstocks and processes, and examine these with a critical eye. We unite universities, researchers, industries and government, with whom we collaborate closely – to greenify the chemical industry, the develop new chemistry for future industry and to educate the next generation of chemists.



Greenifying Chemistry

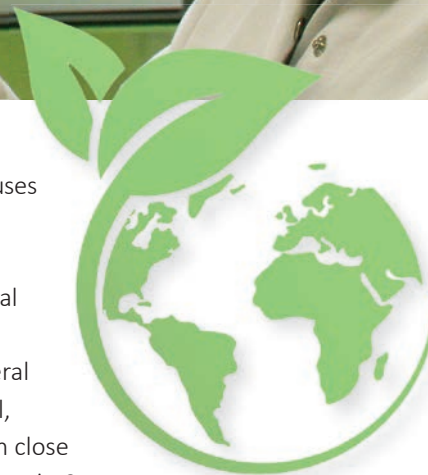
Despite the diversity of projects that are being launched and rounded off, our focus remains unchanged: To greenifying chemistry to accelerate the transition towards Sustainable Advanced Materials made by Circular Process and Energy Technologies as a basis for a Safe and Healthy society and a waste-free Circular economy.



Our Research

Our research programme is constructed around the principle of developing alternative and – above all – greener methods for the chemical industry. Our ultimate goal is to close ‘the product loop’ and, by doing so, transform it into a fully recyclable model.

Therefore, part of our programme focuses on the conversion of various types of feedstocks for the bulk chemicals industry, including biomass, CO₂, natural gases (such as methane) and plastic waste. ARC CBBC conducts both bilateral and multilateral projects. Our bilateral, topic-specific research is carried out in close collaboration with our industrial partners’ R&D divisions. Together, we aim to improve specific industrial processes where they matter most. The multilateral projects are set-up in collaboration with all industrial partners focusing collectively on one specific common goal. These have the



potential to yield ground-breaking results in the race towards a green and circular industry.

The results of our work regularly find their way to a huge diversity of scientific journals. Several of these publications were even featured on their covers! A comprehensive overview of our publications can be found on our website. [↗](#)

Photo: Researchers Floor Brzesowky (left) and Sofie Ferwerda (both Utrecht University) shining light on IR cell in lab



1. The energy transition

Our energy supply has always been based on the use of fossil fuels, and moving away from this is far easier said than done. A key element in the transition to clean energy is the switch to electricity as an energy source. We are constantly learning more about how to generate this more sustainably. This process, called electrification, is becoming increasingly visible all around us, as illustrated by the increasing popularity of electrically powered cars and heat pumps. Switching to electricity as an energy source is also of crucial importance in the chemical industry. Of course, this does require the redesign of the customary chemical production processes. Our chemists are eager to set to work on reinventing these processes and making the switch to electrically powered production processes. One of these chemists is Alexandra Matei, whose work is presented on Page 14.

2. The materials transition

What is better than the sustainable manufacture of new products? Ensuring that the products we are already using are safe and sustainable by design! We can take a significant step in the right direction by ensuring that products last longer or can be augmented with additional functions, so that no new products need to be made. Redesigned materials also lead to energy savings. Can we create more efficient catalysts, for example, so that chemical reactions are less energy-consuming? And can we simultaneously ensure that the production of these catalysts no longer require the use of expensive and hazardous materials? Take a look at Jens Tolboom's research on Page 4 for a fine example of this.



3. The feedstock transition

Many of our products have their origins in non-renewable feedstocks. Plastics, cleaning agents and other products made by the chemical industry are all examples of this. By replacing these non-renewable, often fossil feedstocks with renewable alternatives, we can take a significant step in the greenification of the chemical industry. Can we use waste, such as plastics, CO₂ and biomass to make our products? The projects led by Işıl Yeşil Gür (Page 9) and Bas den Hartigh (Page 10) are both excellent examples of this transition.

Işıl Yeşil Gür (TU/e)

Over the past few decades, we have become highly dependent on plastics due to their outstanding practical properties and durability. However, these same qualities also pose a significant drawback. Once discarded, plastic products become a major societal concern – yet plastic waste, if repurposed as a raw material, could potentially help meet the demand for a wide range of products. Couldn't we simply use plastic waste – currently a major problem – as a feedstock instead of fossil raw materials like petroleum? The real challenge lies in finding a way to turn that plastic into reusable materials.



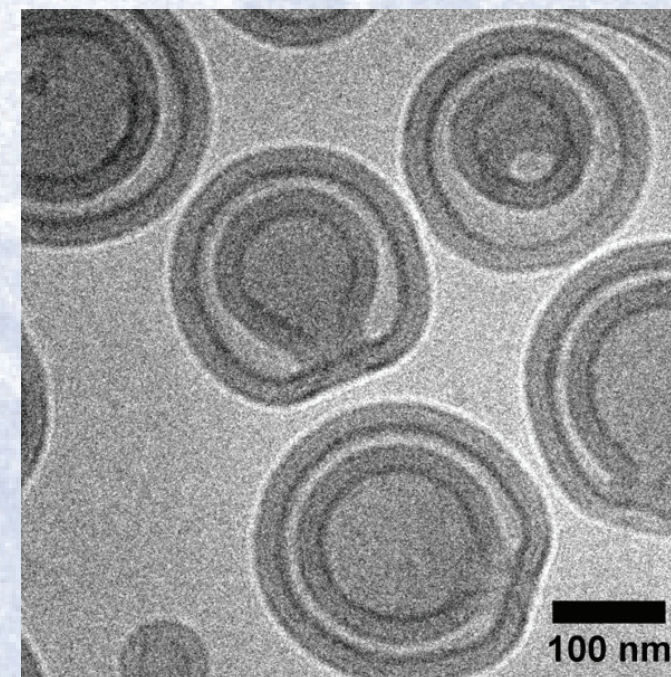
The answer to this question depends entirely on the type of plastic, and the desired end product. Işıl Yeşil Gür is working at Eindhoven University of Technology on polyolefins – a group of thermoplastics widely used in food packaging, among other applications. She is developing tiny machines called nanoreactors that convert polyolefin waste into usable materials. The end products she aims to produce through her research include new raw materials for plastic manufacturing as well as fuels.

The nanoreactors created by Işıl are bowl-shaped, with a central cavity. In the very centre of the nanoreactor, the

plastic waste is broken down and converted into products that can be reused.

The cavity of the nanoreactor houses a specially developed catalyst, which is ultimately responsible for carrying out the entire process. This catalyst is being developed by Victor Drozhzhin, a researcher at ARC CBBC, as part of Işıl's multilateral project 'New Chemistry for a Sustainable Future'. It's a great example of how our researchers collaborate to develop innovative, sustainable solutions for the future!

Photo below: Işıl's nanoreactors



Bas den Hartigh (UU)

If any example truly embodies the modern mantra 'from waste to resource,' it is the use of CO₂ as a feedstock. Carbon dioxide is arguably the biggest threat to our climate today – one we often wish would simply disappear. But if we could repurpose CO₂ as a feedstock, we could achieve two goals at once: lowering CO₂ levels while decreasing our manufacturing industry's reliance on fossil resources. And this isn't just wishful thinking – the process is already taking shape in laboratories, such as the one at Utrecht University where Bas den Hartigh is working.


CO₂ is a highly stable molecule, perfectly content in its current form. However, while this might seem like an advantage, it actually makes CO₂ much harder to process. Processing CO₂ requires significant amounts of energy, and consequently, money and raw materials. To transform CO₂ into usable feedstocks, catalysts are often employed. Given the wide



variety of catalyst types and sizes, their design is crucial to their efficiency in converting CO₂.

Bas is working on what is referred to as the 'hydrogenation' of CO₂, which means that he is adding hydrogen atoms to CO₂ molecules. This process is actually the opposite of what happens when fuels like natural gas and petrol are burned, where the hydrogen atoms are stripped from the fuels. Hydrogenation essentially reverses the combustion of these fuels, resulting in new fuels or building blocks for creating other synthetic products.

In his research, Bas is investigating how hydrogenation of CO₂ occurs on the catalyst. He is examining the role of the different building blocks of the catalyst, cobalt and titanium oxide, and how they support each other. With a better understanding of this process, we can create more effective catalysts and bring the application of CO₂ conversion in society one step closer. Bas also presented his research at the 2024 ChemistryNL Future Festival.

 ARC CBBC at the ChemistryNL Future Festival

Outreach - events

You have probably encountered ARC CBBC at various places last year. Aside from the events hosted by our organization, where we offered various researchers and organizations an opportunity to present themselves, we participated in several external events. We were also visible in various media! You can find this on page 12.



Matteo Monai at NWO CHAINS



Rohit Raj at ChemistryNL Future Festival
photo: Bram Becks



ARC CBBC Community Event
photo: Marijke Badings



Daan Groefsema at Natural Gas Conversion Symposium



ARC CBBC visit to BASF Ludwigshafen



Nathália Tavares Costa at N3C

Outreach - in the media

Besides the ARC CBBC special that C2W magazine published, our people appeared in various media outlets. Find out more about them and their research!



In 2024, C2W magazine published a special edition about ARC CBBC. An edition with all you need to know about ARC CBBC, and full of stories from our researchers! [↗](#)



George Hermens (alumnus) in VVVF "George Hermens, Researcher bij AkzoNobel" [↗](#)



Thomas Freese (RUG) in Nature Spotlight "The Trials and Triumphs of Sustainable Science" [↗](#)

Thomas Freese (RUG) in C2W "Improving Sustainability Starts in your Own Lab" [↗](#)



Daniela Rodrigues Silva (VU) in C2W Magazine: "Steric effects often overlooked" [↗](#)



Nong Artrith (UU) in Eye Openers "Harnessing AI for a Greener Future" [↗](#)



Maurits de Roo (RUG) in Eye Openers "Doing Chemical Reactions using Electricity instead of Chemicals" [↗](#)



Bert Weckhuysen (UU) in de Volkskrant "Klimaatneutraal staal smelten, cement maken en plastic produceren? Dat kan, maar je moet er wel wat voor over hebben" [↗](#)

Bert Weckhuysen (UU) in MO* Magazine "Nederlands onderzoek in Nature: 'Fossielvrije raffinaderij mogelijk in 2050'" [↗](#)

Education



The ARC CBBC education program plays a crucial role in our consortium's strategy to equip the next generations of scientists and employees. Our main goal is to undergo four years of intensive scientific development, but at the same time foster collaborative skills, soft skills, and expertise exchange, complemented by active industry involvement.

 [ARC CBBC Summer School 2024](#)

Top photo: Interdisciplinary training from Jessica Oudenampsen during ARC CBBC Summer School

Round photo: Sailing on a Frisian skûtsje during ARC CBBC Summer School

A significant component in our education programme is our annual Summer School. During the Summer School, for three enriching days, we bonded over a shared mission through community-building activities, insightful scientific lectures, practical business cases, engaging workshops, and an unforgettable boat trip. As part of our commitment to sustainable industry, we invite our partners to host a day during this three-day event. This year, we were honoured to visit BASF in Heerenveen. Rob Gosselink and Gerald Metselaar from BASF curated an active program featuring four business cases centred on sustainability. During the summer school, our students presented their innovative perspectives on specific topics within BASF's sustainability strategy to BASF's experts in Sustainability and R&D.



ARC CBBC wants its PhD students to be involved in today's society; to be able to convey the importance of their research to the general public, stakeholders, industry and politics.

We believe that talent manifests itself in various skills. Being an excellent scientist is one, but so is the ability to take on a leadership role or enthuse a lay audience for your work. All those qualities help shape the researchers of today. As such, ARC CBBC offers workshops that focus on helping students improve their soft skills. A prime example of this is a presentation training course that is given each year. This popular course is well-received, and the students who take part in it are often invited to present their research at events organised by ARC CBBC.

Alexandra Matei (RUG)

Sometimes, you know instantly what you want to achieve. Whether it's a goal, a destination or a product: you often know exactly what you want. The same applies to the greenification of society. We know what we want to achieve. We want to reduce our CO₂ emissions, produce less waste, and turn any waste nevertheless produced into reusable feedstock – and so on.

However, the chemical processes leading to these solutions are often highly complex, expensive, dependent on rare metals, or energy intensive. Our goals may be crystal-clear, but the journey to achieve them is often fraught with many obstacles. Can't we find a way around them? That is exactly what Alexandra Matei is seeking in her research at the University of Groningen.

Alexandra is not so much working on a product, but a chemical production process. She is developing a

methodology designed to minimise adverse effects: she uses inexpensive catalysts that are readily available in large quantities and strives to use water or air as benign parts of the chemical reactions. Many chemical processes rely on complex additives, so shifting to water or air would significantly reduce both costs and effort. In addition, she uses electricity or light as an energy source for her reaction, both of which are derived from sustainable means. She also endeavours to ensure that her process generates the least possible amount of waste. This may sound like an ideal world, but Alexandra is working to turn it into reality. By focusing

on the process itself rather than the feedstock, she can concentrate fully on enhancing its sustainability without being constrained by peripheral requirements. Obviously, the next step is to determine the applications for which this process can be used.

You could compare Alexandra's work to designing a new route through a complex landscape. The starting point of her journey is not the most important; instead, the focus is on the journey. Once the route is mapped out, hopefully, a new path will unfold on the route to a more sustainable world.



Jonas Gans (DIFFER)

DIFFER, the Dutch Institute for Fundamental Energy Research, is one of ARC CBBC's newest partners. This institute specialises in fusion energy and chemical energy research, making it a brilliant asset to the consortium. The first official project at DIFFER has been in progress for some time now: PhD candidate Jonas Gans is conducting research at DIFFER into the production of nitrogen oxides (NO_x i.e. NO and NO_2) through plasma technology.

Key uses of NO_x are the manufacture of nitric acid an indispensable component for the production of polyurethane, a widely used synthetic material, and ammonium nitrate, an effective and common fertiliser. However, the production of NO_x is contingent upon an industrial process that is among the most polluting worldwide, as it requires the use of hydrogen. The 'grey'



hydrogen used in this process is today still predominantly derived from natural gas, releasing CO_2 . Although numerous initiatives have already been established to produce 'green' hydrogen, the demand for this green hydrogen far exceeds the supply, so we, for now, unfortunately remain dependent on grey hydrogen.

Redesigning the NO_x production process to eliminate the need for hydrogen altogether would be a great help – and this is exactly what Jonas aims to do. He is examining the possibility of heating nitrogen and oxygen, both of which are readily available in our atmosphere, to such an extent that they will 'merge into' NO_x , as it were. That would save a lot of fossil feedstocks and prevent a great deal of the accompanying emissions. There is, however, one obstacle: the optimal temperature for this process is around 3200°C . Obviously, this is very difficult to achieve!

This is exactly where the expertise at DIFFER proves invaluable. DIFFER's plasma technology can achieve temperatures of up to 9000°C – in an energy efficient way, thus not only providing the perfect environment for the reaction but also a perspective for the temperature problem. Jonas is using this plasma reactor in his research; he is converting nitrogen and oxygen gases into plasma, a state in which atoms are separated and form a kind of 'atomic cloud' which will be controlled in a way that the atoms can recombine to form NO_x .

The challenge, however, is the competitiveness against established fossil-based production ways on an energy level. Therefore, one aspect of Jonas Research is focused on energy costs, because the process will need to become a lot more energy-efficient to be economically viable. If Jonas' research proves a success, this would give us a new, sustainable production option.

Awards, Grants and other Honours

Click them
to learn
more!



Bert Weckhuysen

- receives Earl L. Muetterties Memorial Lectureship in Chemistry
- receives Honorary Doctorate at Ghent University Belgium
- receives Incubator Grant
- receives Karl Wamsler Innovation Award 2024



Jan van Hest

- receives ERC Advanced Grant



Daan Groefsema

- Receives Best Oral Presentation Award at Natural Gas Conversion Symposium



Adri Minnaard

- receives Ammodo Science Award



Nikolay Kosinov

- receives ERC Consolidator Grant
- receives NWO ENW-M1 Grant



Xinwei Ye

- awarded Embassy Science Fellowship



Ben Feringa

- becomes Member of the American Philosophical Society



Hans Kuipers

- receives Knighthood in the Order of the Netherlands Lion



Bas de Bruin

- becomes Scientific Director of the Van 't Hoff Institute for Molecular Sciences



Jens Tolboom

- receives Best Young Scientist Contribution at ETCC



Petra de Jongh

- receives ERC Advanced Grant



Wiebe de Vos

- Inauguration as professor



Ina Vollmer

- receives ERC Starting Grant
- receives Incubator Grant
- receives NWO Demonstrator Grant



Marina Karsakova

- receives Poster Prize at ARC CBBC Community Event
- receives Poster Prize at the Münster Symposium on Intelligent Matter

Our People

Executive Board (EB) Members

Prof. Bert Weckhuysen (Scientific Director) – Utrecht University
Prof. Ben Feringa (Chair) – University of Groningen
Prof. Hans Kuipers – Eindhoven University of Technology
Ir. André van Linden – AkzoNobel
Dr Peter Berben – BASF
Dr Rob Gosselink – BASF
Dr Evren Ünsal – Shell

The EB members are supported by the following knowledge experts:

Dr Jitte Flapper – AkzoNobel
Dr ir. Sander van Bavel – Shell
Manon van Asselt – NWO

The following members have left the EB in 2024:

Dr Peter Berben – BASF

The following member has joined the EB in 2024:

Dr Rob Gosselink – BASF



Supervisory Board (SB) Members

Marinke Wijngaard MSc – Chair
Prof. Anton Pijpers – Utrecht University
Dr Katrin Frieze – BASF
Dr David Williams- AkzoNobel
Prof. Joost Frenken- University of Groningen
Prof. Dr Silvia Lenaerts – Eindhoven University of Technology
Prof. Rolf van Benthem – Shell

The SB members are supported by the following observers:

Ir. Jacqueline Vaessen – ChemistryNL
Manon van Asselt MSc – NWO

The following members have joined the SB in 2024:

Rolf van Benthem – Shell

Scientific Advisory Board (SAB) Members

Prof. Matthias Beller, Chair – Leibniz-Institut für Katalyse, Germany
Prof. Markus Antonietti – Max-Planck Institute of Colloids and Interfaces, Germany
Prof. Christophe Copéret – ETH Zürich, Switzerland
Prof. Tanja Cuk – University of California at Berkeley, CA, USA
Prof. Rodney O. Fox – Iowa State University, USA
Prof. Joseph Keddie – University of Surrey, UK
Prof. Martin Möller – Leibniz Institute for Interactive Materials, Germany
Prof. Ferdi Schüth – Max-Planck-Institut für Kohlenforschung, Germany
Prof. Timothy Swager – Massachusetts Institute of Technology, USA

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Prof. Guy Marin, Deputy Chair – Ghent University, Belgium
 Prof. Beatriz Roldan – Fritz Haber Institute of the Max Planck Society, Germany
 Prof. Helma Wennemers – ETH Zürich, Switzerland
 Prof. Unni Olsbye – University of Oslo, Norway
 Prof. Raffaella Buonsanti – EPFL, Switzerland
 Prof. Marc-Olivier Coppens – University College London, UK
 Dr Hélène Olivier-Bourbigou – IFP Energies Nouvelles, France

The following members have joined the SAB in 2024:

Prof. Raffaella Buonsanti – EPFL, Switzerland
 Prof. Marc-Olivier Coppens – University College London, UK
 Dr Hélène Olivier-Bourbigou – IFP Energies Nouvelles, France

The following member has left the SAB in 2024:

Prof. Guy Marin, Deputy Chair – Ghent University, Belgium



Photo: David de Wied building

Photo right: Lecture Ellard Hooijveld at the ARC CBBC Community Event

Photo page 18: Vening Meinesz building

Members

Prof. Adri Minnaard – Groningen University
 Prof. Albert Schenning – Eindhoven University of Technology
 Prof. Alfons van Blaaderen – Utrecht University
 Prof. Atsushi Urakawa – Delft University of Technology
 Prof. Bas de Bruin – University of Amsterdam
 Prof. Ben Feringa – University of Groningen
 Prof. Bert Weckhuysen – Utrecht University
 Dr Catarina de Carvalho Esteves – Eindhoven University of Technology
 Prof. Detlef Lohse – University of Twente
 Prof. Emiel Hensen – Eindhoven University of Technology
 Prof. Erik Garnett – University of Amsterdam
 Prof. Evgeny Pidko – Delft University of Technology
 Prof. Guido Mul – University of Twente
 Prof. Hans Kuipers – Eindhoven University of Technology
 Prof. Jan van Hest – Eindhoven University of Technology
 Prof. Jasper van der Gucht – Wageningen University & Research
 Prof. Joost Reek – University of Amsterdam
 Prof. Kitty Nijmeijer – Eindhoven University of Technology
 Prof. Marc Koper – Leiden University
 Prof. Matthias Bickelhaupt – Vrije Universiteit Amsterdam
 Prof. Moniek Tromp – Groningen University
 Dr Monique van der Veen – Delft University of Technology
 Prof. Nathalie Katsonis – University of Twente
 Dr Nong Artrith – Utrecht University
 Prof. Peter Bolhuis – University of Amsterdam
 Prof. Petra de Jongh – Utrecht University
 Prof. Pieter Bruijninx – Utrecht University
 Prof. René Janssen – Eindhoven University of Technology
 Prof. Richard van de Sanden – DIFFER
 Prof. Ruud van Ommen – Delft University of Technology
 Prof. Sijbren Otto – Groningen University
 Prof. Syuzanna Harutyunyan – Groningen University
 Prof. Thijs Vlugt – Delft University of Technology
 Prof. Wesley Browne – Groningen University
 Prof. Wiebe de Vos – University of Twente

Assistant Professors

Dr Matteo Monai – Utrecht University
 Dr Eline Hutter – Utrecht University
 Dr Ina Vollmer – Utrecht University
 Dr Michael Lerch – University of Groningen
 Dr Sebastian Beil – University of Groningen
 Dr Nikolay Kosinov – Eindhoven University of Technology
 Dr Marta Costa Figueiredo – Eindhoven University of Technology

The following Assistant Professor has left in 2024:

Dr Sebastian Beil – University of Groningen

Technicians

Ing. Hannie van Berlo- van den Broek – Utrecht University
 Dr Ramon Oord – Utrecht University
 Dr Peter de Peinder – Utrecht University
 Ing. Larry de Graaf – Eindhoven University of Technology
 Ir. Brahim Mezari – Eindhoven University of Technology
 Lotte Stindt MSc – University of Groningen
 Dr Alexander Ryabchun – University of Groningen

The following technicians have joined in 2024:

Lotte Stindt – University of Groningen
 Dr Alexander Ryabchun – University of Groningen

ARC CBBC Support Office

The ARC CBBC Support Office is hosted by the coordinating partner, Utrecht University.

Anita ter Haar – Financial Controller
 drs. Hannah Thuijs – Consortium Manager
 Dr Esther Groeneveld – Consortium Manager
 Anita den Heijer – Office Manager
 Masja Spijckstra – Project Coordinator
 Jeroen Meijer MSc – Communication Advisor
 Marijke Badings – Communication Officer / Graphic designer



Our New Researchers

The following PhD candidates and postdoctoral researchers have joined ARC CBBC in 2024.



Paul Wienecke
(RUG; PD)



Marika Di Berto Mancini
(RUG; PD)



Stefan Bouts
(TU/e; PhD)



Aleksandr Bashkatov
(UT; PD)



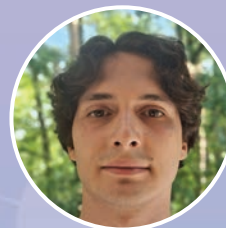
Yixin Zhang
(UT; PD)



Rik Bennis
(UvA; PhD)



Alessio Baldelli
(TU/e; PhD)



Marzio Saccinto
(UT; PhD)



David Vesseur
(UvA; PD)



Kalani Ostermeijer
(TUD; PhD)



Arvid Beeuwkes
(UU; PhD)



Batuhan Özyürek
(RUG; PhD)



Martijn Hoving (RUG; PhD)



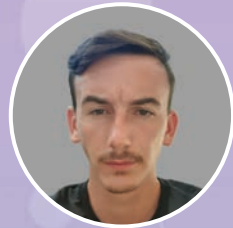
Ferit Begar (RUG; PhD)



Devanshu Sajwan (UU; PhD)

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Our Alumni

The following people have successfully defended their thesis in 2024, and can now call themselves ARC CBBC alumni!



Johan Bootsma (UvA)



Francesco Mattarozzi (UU)



Marie Brands (UvA)



Kordula Schnabl (UU)



Felix de Zwart (UvA)



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Kelly Brouwer (UU)



Lotte Metz (UvA)



Harith Gurunarayanan (UU)



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